### Project Title:

# Density functional theory studies on the mechanisms of transition metal mediated chemical transformations

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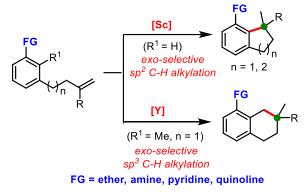
### Laboratory at RIKEN:

RIKEN Center for Sustainable Resource Science, Advanced Catalysis Research Group
RIKEN Cluster for Pioneering Research, Organometallic Chemistry Laboratory

## 1. Background and purpose of the project, relationship of the project with other projects

Benzo-fused carbocycles such as indanes and tetralins, especially those bearing an all-carbon quaternary stereocenter, are important structural motifs in a broad spectrum of natural products, bioactive molecules and functional materials. Therefore, the development of efficient and selective methods for the synthesis of indane and tetralin derivatives having an all-carbon quaternary stereocenter is of much interest and importance. In principle, the Markovnikovselective intramolecular C-H addition of an aromatic unit to 1,1-disubstituted alkenes would be the most atom-economical and straightforward route for the construction of indane and tetralin skeletons bearing an all-carbon quaternary center. However, this approach has met with only limited success to date because of the lack of suitable catalysts.

Recently, half-sandwich rare-earth catalysts have been found that could serve as a unique platform for various chemical transformations, including C-H alkylation with alkenes. In the course of these studies, we became interested in exploring the potential of rare-earth catalysts for the construction of important carbocycles through intramolecular C-H alkylation with alkenes. Herein, we report for the first time the annulation of a wide range of functionalized aromatic substrates with 1,1-disubstituted alkenes via rareearth-catalyzed regiospecific  $C(sp^2)$ -H and benzylic  $C(sp^3)$ -H activation (**Fig. 1**). The reaction took place in an *exo*-selective fashion, constituting an efficient and selective route for the synthesis of a new family of indane and tetralin derivatives with an all-carbon quaternary stereocenter which were difficult to access previously. To have a better understand the mechanism and *exo*-selectivity of this reaction, DFT calculations were performed.



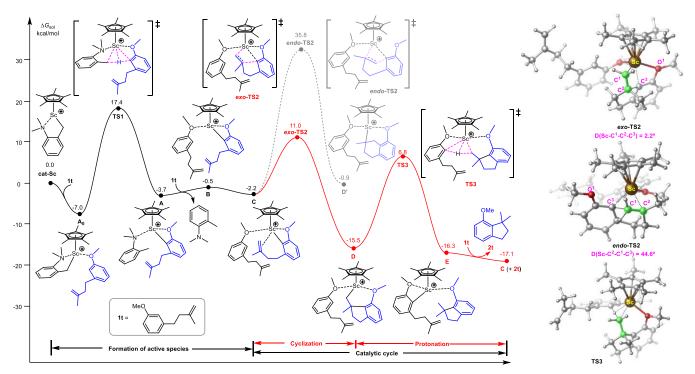
**Fig. 1.** Rare-earth-catalyzed *exo*-selective annulation affording an all-carbon quanternary stereocenter.

### 2. Specific usage status of the system and calculation method

In the last FY2022, 40% computing resources were used for my Quick User project. All calculations were performed by Gaussian 16 software together with DFT methods, such as B3LYP, B3LYP-D3, M06, M06L, M062X, B3PW91 and so on.

### 3. Result

As shown in Fig. 2, the coordination of the methoxy group of **1t** to the Sc atom in the cationic species  $[(C_5Me_5)Sc(CH_2C_6H_4NMe_2 - o)]^+$  (**cat-Sc**) generated by the reaction of **Sc-2** with  $[Ph_3C][B(C_6F_5)_4]^{[10]}$  gave a more stable intermediate **A**<sub>0</sub>, which after C-H activation at the C2 position of the coordinated anisole unit through *s*-bond metathesis with the benzyl scandium species yielded the anisyl species **A**. This process required overcoming an energy barrier of 24.4 kcal/mol. Replacement of the  $\sigma$ -Me<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>Me ligand in **A** with another molecule of **1t** could afford **B**, which may subsequently be transformed to **C** through the intramolecular coordination of the C=C bond in the anisyl group to the metal center. The insertion of the coordinated C=C bond into the anisyl Sc-C bond (cyclization) in **C** could easily occur in an *exo*-selective fashion by overcoming an energy barrier of 13.2 kcal/mol via transition state *exo*-**TS2**, thus forming a five-membered ring with a quaternary stereocenter (**D**). In comparison, the *endo*-selective C=C insertion in **C** *via endo*-**TS2** to give a sixmembered ring product **D'** would require a much higher activation energy (38.0 kcal/mol).



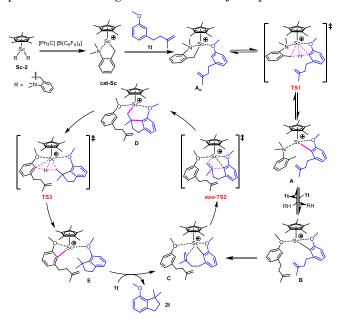
**Fig. 2**. Energy profile for the scenarios of the intramolecular C–H annulation of an alkenyl-substituted anisole (**1t**) catalyzed by **Sc-2**. The relative free energies ( $\Delta G^{\ddagger}$ , kcal/mol) in solution were calculated at the level of M06/6-311+G(d,p)&SDD (SMD, toluene)//B3PW91/6-31G(d)&SDD.

The large difference in activation energy between **exoTS2** and **endoTS2** was caused by both steric and electronic factors. In **endoTS2**, there was a large distortion in the core four-membered ring (Sc-C2-C1-C3) as shown by the large distorted dihedral angle  $D(Sc-C2-C1-C3) = 44.6^{\circ}$ , and the coordination interaction between the Sc atom and the O1 atom in the anisyl unit no longer existed. However, in **exoTS2** the four-membered ring (Sc-C1-C2) was almost planar (D(Sc-C1-C2-C3) = 2.2°) without

showing a significant distortion.<sup>[18]</sup> The C2 $\Box$ H activation of the coordinated anisole unit by the Sc–C alkyl bond in **D** could yield **E**, which upon coordination of another molecule of **1t** finally released the *exo*-annulation product **2t** and regenerated the catalytic species **C**. The transformation of **D** to **E** *via* transition state **TS3** ( $\Delta G^{\ddagger} = 22.3$  kcal/mol) represents the rate-determining step in the catalytic cycle, in agreement with the KIE observed experimentally.

### 4. Conclusion

On the basis of DFT calculations, a possible reaction mechanism for the exo-selective annulation was revealed (Fig. 3). The coordination of the anisole unit of 1t to the Sc atom in the cationic species cat-Sc would give an adduct complex like A<sub>0</sub>, which may undergo either C2-H activation through s-bond metathesis to afford the corresponding anisyl species A. The replacement of the *o*-Me<sub>2</sub>NC<sub>6</sub>H<sub>4</sub>Me ligand in A with another molecule of 1t through ether coordination to the metal center could afford B, which may subsequently be transformed to C. The exoselective intramolecular insertion of the coordinated C=C bond into the anisyl Sc–C bond in **C** would afford an indane skeleton with an all-carbon quaternary stereocenter like D. The C2-H activation of the coordinated anisole unit by the Sc–C alkyl bond in **D** may give E, which upon coordination of another molecule of 1t would finally release the indane product **2t** and regenerate the catalytic species **C**.



**Fig. 3**. Possible reaction mechanism for *exo*-selective annulation of **1t** by **Sc-2** on the basis of DFT calculations.

#### 5. Schedule and prospect for the future

In the following FY2023, I plan to continue collaboration with Prof. Zhaomin Hou (Organometallic Chemistry Laboratory & Advanced Catalysis Research Group, RIKEN) to investigate the related mechanisms of the metal-mediated homogeneous chemical reactions, including small molecule activation, olefin polymerization, C-H alkylation and so on. The mechanism will also be investigated by DFT calculations. Therefore, I want to get the continuous support from RIKEN Supercomputer System in the future.

### Usage Report for Fiscal Year 2022 Fiscal Year 2022 List of Publications Resulting from the Use of the supercomputer

 Aniket Mishra, Ping Wu, Xuefeng Cong, Masayoshi Nishiura, <u>Gen Luo\*</u>, Zhaomin Hou\*. Exo-Selective Intramolecular C–H Alkylation with 1,1-Disubstituted Alkenes by Rare-Earth Catalysts: Construction of Indanes and Tetralins with an All-Carbon Quaternary Center. *ACS Catalysis* 2022, *12*, 12973–12983. (Publication Date: Oct. 11th, 2022; Impact Factor: 13.700)